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núcleos de ^{12}C em 1 g de carbono é $6,02 \times 10^{23}$ núcleos mol ^{12}C mol n 1 g n $6,02 \times 10^{23}$ ^{12}C 5 ^{12}C 2 ^{12}C núcleos g o número de núcleos de ^{14}C radioativo é então igual a razão $1,3 \times 10^{12}$ vezes n ou seja $5,02 \times 10^{22}$

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web quando um decaimento beta cria um positron ele é chamado de decaimento beta positivo e quando ele cria um elétron ele é chamado de decaimento beta negativo decaimento gama desenho que mostra o decaimento gama