

REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

H *hydrogen*
S *sulphur*
Cl *chlorine*
C *carbon*
N *nitrogen*
Na *sodium*
Pb *lead*
Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*
phosphorus *P*
fluorine *F*
magnesium *Mg*
potassium *K*
calcium *Ca*
zinc *Zn*
iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79** copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5** calcium (Ca) **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Lead	82	207	82	125	82
Argon	18	58	18	40	18
Barium	56	137	56	61	56
Tin	50	112	50	62	50

6. Match the unit with its description:

F 1. amount of substance
C 2. density
B 3. mass
D 4. molar mass
A 5. energy
E 6. volume

A. joule
B. g
C. g/mL
D. g/mole
E. L
F. mole

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David Baud



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Table of Contents Basic Chemistry And Basic Applications Guide Answers

1. Understanding the eBook Basic Chemistry And Basic Applications Guide Answers
 - The Rise of Digital Reading Basic Chemistry And Basic Applications Guide Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Basic Chemistry And Basic Applications Guide Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Basic Chemistry And Basic Applications Guide Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Basic Chemistry And Basic Applications Guide Answers
 - Personalized Recommendations

- Basic Chemistry And Basic Applications Guide Answers User Reviews and Ratings
- Basic Chemistry And Basic Applications Guide Answers and Bestseller Lists
- 5. Accessing Basic Chemistry And Basic Applications Guide Answers Free and Paid eBooks
 - Basic Chemistry And Basic Applications Guide Answers Public Domain eBooks
 - Basic Chemistry And Basic Applications Guide Answers eBook Subscription Services
 - Basic Chemistry And Basic Applications Guide Answers Budget-Friendly Options
- 6. Navigating Basic Chemistry And Basic Applications Guide Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Basic Chemistry And Basic Applications Guide Answers Compatibility with Devices
 - Basic Chemistry And Basic Applications Guide Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Basic Chemistry And Basic Applications Guide Answers
 - Highlighting and Note-Taking Basic Chemistry And Basic Applications Guide Answers
 - Interactive Elements Basic Chemistry And Basic Applications Guide Answers
- 8. Staying Engaged with Basic Chemistry And Basic Applications Guide Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Basic Chemistry And Basic Applications Guide Answers
- 9. Balancing eBooks and Physical Books Basic Chemistry And Basic Applications Guide Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Basic Chemistry And Basic Applications Guide Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Basic Chemistry And Basic Applications Guide Answers
 - Setting Reading Goals Basic Chemistry And Basic Applications Guide Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Basic Chemistry And Basic Applications Guide Answers
 - Fact-Checking eBook Content of Basic Chemistry And Basic Applications Guide Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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