

Additivity of Heats of Reaction: Hess's Law

In this experiment you will use a Styrofoam cup calorimeter to measure the heat released by three reactions/physical changes. One of them is the same as the sum of the other two. Therefore, according to Hess's Law of Constant Heat Summation (or just Hess's Law), the heat of reaction of the one reaction should be equal to the sum of the heats of reaction for the other two. The primary objective of this experiment is to confirm this law.

The reactions we will use in this experiment are:

- (1) Solid sodium hydroxide dissolves in water to form an aqueous solution of ions.



- (2) Solid sodium hydroxide reacts with an aqueous solution of hydrochloric acid to form water and leave an aqueous solution of sodium chloride.



- (3) Aqueous solutions of sodium hydroxide and hydrochloric acid react to form water and leave an aqueous solution of sodium chloride.



You will use a Styrofoam cup (in a beaker) as a calorimeter, as shown in Figure 1. For purposes of this experiment, you may assume that the heat loss to the calorimeter and the surrounding air is negligible. Even if heat is lost to either of these it is a fairly constant factor in each part of the experiment and has little effect on the final results.



Figure 1

Additivity Of Heats Of Reaction Lab Answers

Tzimopoulo



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This book delves into Additivity Of Heats Of Reaction Lab Answers. Additivity Of Heats Of Reaction Lab Answers is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Additivity Of Heats Of Reaction Lab Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Additivity Of Heats Of Reaction Lab Answers
- Chapter 2: Essential Elements of Additivity Of Heats Of Reaction Lab Answers
- Chapter 3: Additivity Of Heats Of Reaction Lab Answers in Everyday Life
- Chapter 4: Additivity Of Heats Of Reaction Lab Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Additivity Of Heats Of Reaction Lab Answers. This chapter will explore what Additivity Of Heats Of Reaction Lab Answers is, why Additivity Of Heats Of Reaction Lab Answers is vital, and how to effectively learn about Additivity Of Heats Of Reaction Lab Answers.
3. In chapter 2, the author will delve into the foundational concepts of Additivity Of Heats Of Reaction Lab Answers. This chapter will elucidate the essential principles that need to be understood to grasp Additivity Of Heats Of Reaction Lab Answers in its entirety.
4. In chapter 3, the author will examine the practical applications of Additivity Of Heats Of Reaction Lab Answers in daily life. This chapter will showcase real-world examples of how Additivity Of Heats Of Reaction Lab Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Additivity Of Heats Of Reaction Lab Answers in specific contexts. This chapter will explore how Additivity Of Heats Of Reaction Lab Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Additivity Of Heats Of Reaction Lab Answers. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Additivity Of Heats Of Reaction Lab Answers.

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