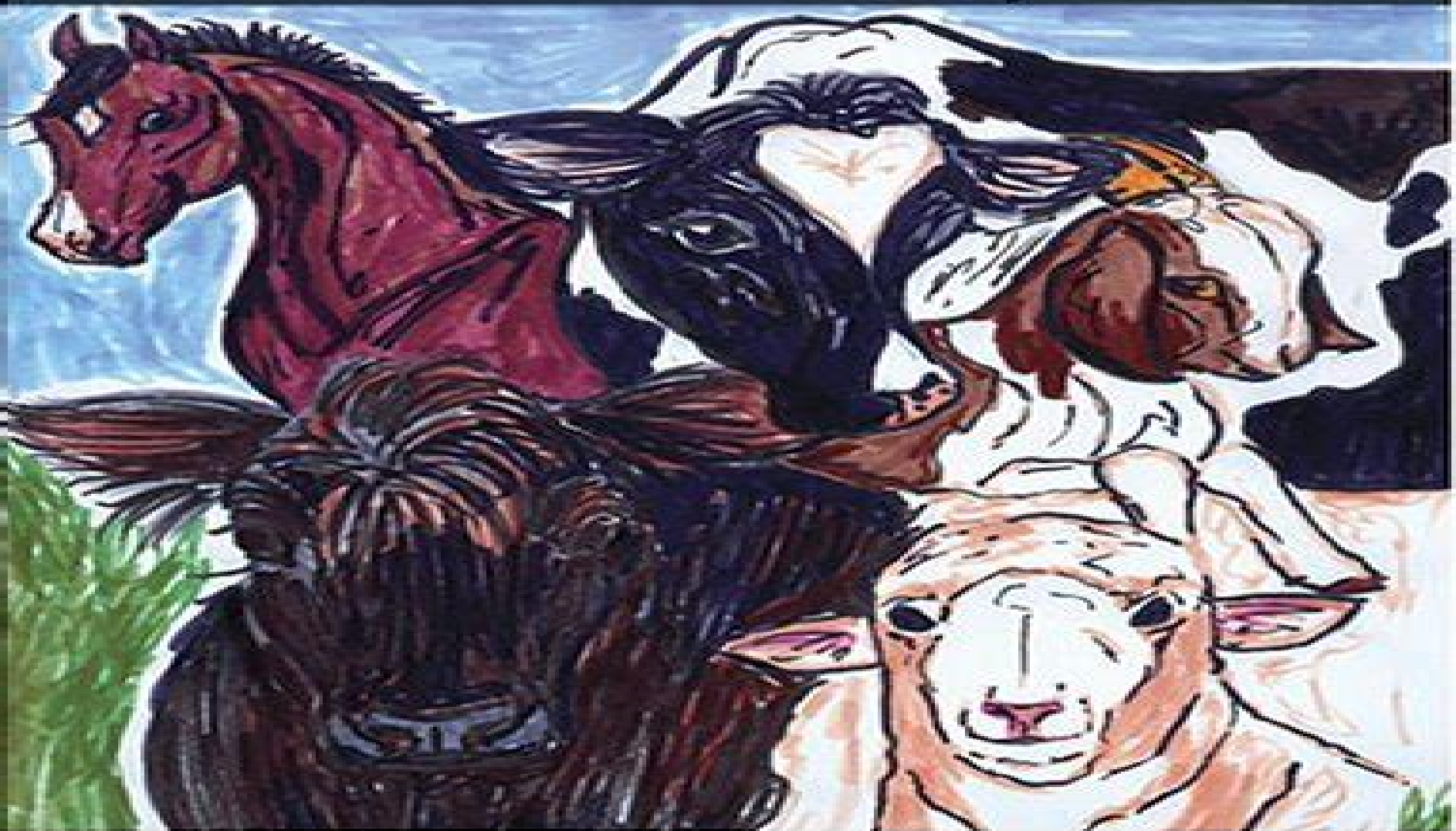


Introduction to Animal Science Laboratory Manual

Nancy A. Kieser



Animal Science Laboratory Manual

Bill V. Able



Animal Science Laboratory Manual:

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Animal Science is designed for post secondary students in an introductory course in a department of animal science The text is designed to prepare students majoring in animal science for more advanced and specialized courses It also provides non animal science majors a comprehensive overview of livestock production Terminology is defined and an extensive glossary is provided The text is extensively illustrated with photographs line drawings charts and tables to expand and illuminate the material A section of color photographs may be used for breed identification It is based on the most up to date information available and is applicable to all areas of the United States *Lab Manual to Accompany Animal Science Biology & Technology* Robert Mikesell, Meecee Baker, 2010-04 The Laboratory Manual is a valuable tool designed to enhance your lab experience Lab activities objectives materials lists step by step procedures illustrations and review questions are commonly found in a Lab Manual *Introductory Animal Science* George Brant, 1985 *Animal Science and Industry Laboratory Manual* Bill V. Able, 1978 *Basic Animal Science* W. T. Berry, Anton Marinus Sorensen, Landon Douglas Wythe, 1978 The Effect of the Introductory Animal Science Laboratory Manual on Student Achievement Patricia A. Riggs, 1983 *Exploring Animal Science: Lab Manual* Frank B. Flanders, 2011-04-01 The Laboratory Manual supplements the text and includes lab exercises

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web compiled by g izquierdo rodríguez page 2 1 directly proportional to the product of their charges $f \propto q_1 q_2$ 2 2 inversely proportional to square of the distance between them $f \propto \frac{1}{r^2}$ therefore $f \propto \frac{q_1 q_2}{r^2}$ coulomb s law in symbols $F = k \frac{q_1 q_2}{r^2}$ coulomb s law the magnitude of the electrostatic force exerted by one point charge q

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charged due to the majority cl ions the solution becomes positively charged due to the stronger mg 2 ions b the solution

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through the sphere is zero but the electric field is not zero anywhere in the sphere

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the test charge to the right from c 2 there is an attractive force on the test charge to the left by referring to coulomb s law □□
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