

# AP CHEMISTRY

Volume 1: Your Second Year Study

## Properties of Solids

**Crystalline Solids**

- Have a regular, repeating arrangement of particles (atoms, ions, or molecules).
- Have sharp melting points.
- Can be cleaved along flat planes.

**Amorphous Solids**

- Do not have a regular, repeating arrangement of particles.
- Do not have sharp melting points.
- Do not cleave along flat planes.

**Intermolecular Forces**

- London Dispersion Forces:** Present in all molecules, but only significant in nonpolar molecules.
- Dipole-Dipole Forces:** Present in polar molecules.
- Hydrogen Bonding:** Present in molecules containing N-H, O-H, or F-H bonds.

### Periodic Table & Ionization Energies

Ionization energy generally increases from left to right and from bottom to top across the periodic table.

### Properties of the Periodic Table

- Metals:** Located on the left side of the table.
- Nonmetals:** Located on the right side of the table.
- Metalloids:** Located between metals and nonmetals.
- Groups:** Vertical columns of elements.
- Periods:** Horizontal rows of elements.

### Electronegativity Table

**Electronegativity increases from bottom-left to top-right.**

## Properties of Liquids

**Intermolecular Forces**

- London Dispersion Forces:** Present in all molecules.
- Dipole-Dipole Forces:** Present in polar molecules.
- Hydrogen Bonding:** Present in molecules containing N-H, O-H, or F-H bonds.

## Properties of Gases

**Ideal Gas Law:**  $PV = nRT$

**Real Gases:** Deviate from ideal behavior at high pressures and low temperatures.

### Properties of Acids & Bases

| Acid                            | Base                |
|---------------------------------|---------------------|
| HCl                             | NaOH                |
| HNO <sub>3</sub>                | KOH                 |
| H <sub>2</sub> SO <sub>4</sub>  | Ca(OH) <sub>2</sub> |
| H <sub>3</sub> PO <sub>4</sub>  | Mg(OH) <sub>2</sub> |
| H <sub>2</sub> CO <sub>3</sub>  | Al(OH) <sub>3</sub> |
| H <sub>2</sub> SiO <sub>3</sub> | Fe(OH) <sub>3</sub> |
| H <sub>2</sub> BO <sub>3</sub>  | Cr(OH) <sub>3</sub> |
| H <sub>2</sub> GeO <sub>3</sub> | Mn(OH) <sub>2</sub> |
| H <sub>2</sub> SnO <sub>3</sub> | Ni(OH) <sub>2</sub> |
| H <sub>2</sub> PbO <sub>3</sub> | Co(OH) <sub>2</sub> |
| H <sub>2</sub> UO <sub>3</sub>  | Ni(OH) <sub>2</sub> |
| H <sub>2</sub> ThO <sub>3</sub> | Co(OH) <sub>2</sub> |
| H <sub>2</sub> PaO <sub>3</sub> | Ni(OH) <sub>2</sub> |
| H <sub>2</sub> AcO <sub>3</sub> | Co(OH) <sub>2</sub> |
| H <sub>2</sub> AsO <sub>3</sub> | Ni(OH) <sub>2</sub> |
| H <sub>2</sub> SbO <sub>3</sub> | Co(OH) <sub>2</sub> |
| H <sub>2</sub> BiO <sub>3</sub> | Ni(OH) <sub>2</sub> |
| H <sub>2</sub> AtO <sub>3</sub> | Co(OH) <sub>2</sub> |
| H <sub>2</sub> FrO <sub>3</sub> | Ni(OH) <sub>2</sub> |
| H <sub>2</sub> BrO <sub>3</sub> | Co(OH) <sub>2</sub> |
| H <sub>2</sub> IO <sub>3</sub>  | Ni(OH) <sub>2</sub> |
| H <sub>2</sub> TeO <sub>3</sub> | Co(OH) <sub>2</sub> |
| H <sub>2</sub> SeO <sub>3</sub> | Ni(OH) <sub>2</sub> |
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**Jiyuan Zhang**



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web penelitian ini dilakukan pada agustus 2019 hingga februari 2020 dengan metode membandingkan hasil dari tiga taraf perlakuan lama waktu inkubasi limbah padat dengan t1 60 hari setelah diangkat dari kolam ipal t2 120 hari setelah diangkat dari kolam ipal t3 180 hari setelah diangkat dari kolam ipal yang diulang sebanyak empat kali

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efisiensi pengolahan air limbah dengan menggunakan persamaan sebagai berikut  $\text{efipal} = \frac{\text{in}}{\text{out}}$

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