

Redox Exercise (Part 2)

Chapter 20 Worksheet: Redox ANSWERS

1. Determine what is oxidized and what is reduced in each reaction. Identify the oxidizing agent and the reducing agent, also.



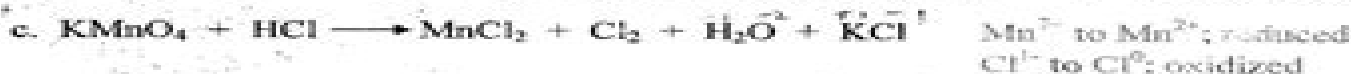
11. Give the oxidation number of each kind of atom or ion.

a. sulfate	b. Sn	c. S^{2-}	d. Fe^{3+}	e. Sn^{4+}	f. nitrate	g. ammonium
2-	0	2-	3+	4+	1-	1+

12. Calculate the oxidation number of chromium in each of the following.

a. Cr_2O_3	b. $\text{Na}_2\text{Cr}_2\text{O}_7$	c. CrSO_4	d. chromate	e. dichromate
3+	6+	2+	6+	6+

13. Use the changes in oxidation numbers to determine which elements are oxidized and which are reduced in these reactions. (Note: it is not necessary to use balanced equations)



14. For each reaction in problem 13, identify the oxidizing agent and reducing agent.

- | | |
|-------------------------------|--------------------------|
| a. oxidizing agent: sulfur | reducing agent: carbon |
| b. oxidizing agent: nitrogen | reducing agent: iodine |
| c. oxidizing agent: manganese | reducing agent: chloride |
| d. oxidizing agent: nitrogen | reducing agent: antimony |

Balancing Redox Equations Chemistry If8766

Marcia L. Gillette, H. Anthony Neidig



Balancing Redox Equations Chemistry If8766:

Principles of Redox Reactions Malika Ammam, 2018-03-05 As a teacher of physical chemistry I noticed that students even in advanced classes have difficulties in understanding the basics of redox chemistry In this Section 3 I attempted to discuss some fundamental principles related to redox processes by focusing on the species that might lose or gain electrons determination of the oxidation numbers or states of atoms in compounds and ways of balancing redox reactions To further clarify the discussed concepts numerous questions and problems with detailed answers are provided Most of these questions are formulated by students like you I believe that this Section 3 would greatly help students with levels varying from high school to advanced university classes

Redox Reactions Explained (General Chemistry Quick Review) E Staff, Learn and review on the go Use Quick Review Science Study Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Perfect study notes for all high school and college students

Balancing Redox Equations Lifelike, 2019 This lesson plan covers how to balance redox equations by inspection through two systematic processes

Balancing Oxidation-Reduction Equations Using the Oxidation Number Method Marcia L. Gillette, H. Anthony Neidig, 1995-03-01 Balancing Organic Redox Equations Harold P. C. Burrell, 1959 **Learning Channel**, 1980

Balancing Oxidation-Reduction Equations Using the Ion-Electron Method Marcia L. Gillette, H. Anthony Neidig, 1995-02-01 A Review of the Methods for Balancing Oxidation-reduction Equations Henry Vivian Stewart, 1962

Redox Mohammed Awad Ali Khalid, 2017-09-06 Redox reactions are central to the major element cycling many cell cycles many chemisorption and physisorption processes trace element mobility from rocks and sediments toward wells aquifers trace element toxicity toward life forms and most remediation schemes including water treatments over the last three decades the field has attracted a lot of scientists and a great deal of researches has been done in redox chemistry This book provides a very broad overview of the state of the art of understanding redox processes which starts with giving a concise introduction that describes the origin historical background and the development of the redox definitions The book is organized into two sections that include ten chapters and introduces in Section 1 generalized electron balance theory and its applications in electrolytic redox systems redox active molecules and its applications in device memory fundamentals and applications of flow batteries and their integration into antidirect current and donor acceptor titrations of displacement and electronic transference Section 2 introduces redox in biological processes including roles of reactive oxygen species in respiration metabolism and regulations and redox in physiological processes as redox sensitive TRP channels TRPA1 and TRPM2 All chapters are written by different authors with the exception of Chapter 1 Introduction This clearly reflects the broad range of topics that have been covered by experts in the field

Sustainable and Functional Redox Chemistry Shinsuke Inagi, 2022-04-29 Mimicking nature's efficiency and sustainability in organic chemistry is a major goal for future

chemists redox reactions are a key element in a variety of fields ranging from synthesis and catalysis to materials chemistry and analytical applications Sustainability is increasingly becoming a consideration in synthesis and functional chemistry and an essential element for the next generation of chemistry in academia and industry This book represents a compilation of the latest advancements in functional redox chemistry and demonstrates its importance in achieving a more sustainable future This book is an ideal companion for any postgraduate students or researchers interested in sustainability in academia and industry

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