



Lab Data

Verify your rate calculation.

Did you report your data to the correct number of significant figures?

Temperature

	Trial 1
Concentration of iodide solution (mM)	100.0
Concentration of thiosulfate solution (mM)	20.1
Concentration of hydrogen peroxide solution (mM)	176.0
Temperature of iodide solution (°C)	4.6
Volume of iodide solution (I^-) used (mL)	10.0
Volume of thiosulfate solution ($S_2O_3^{2-}$) used (mL)	1.0
Volume of hydrogen peroxide solution (H_2O_2) used (mL)	10.0
Time (s)	27.6
Observations	Solution turned blue in 27.6 seconds when
Initial concentration of iodide in reaction (mM)	47.6
Initial concentration of thiosulfate in reaction (mM)	0.96
Initial concentration of hydrogen peroxide in reaction (mM)	83.8
Initial Rate (mM/s)	

Answers To Iodine Clock Reaction Lab

**Ralph Petrucci, William
Harwood, Geoffrey Herring**



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