

Guided Notes: Matter and Chemistry

Introduction

1. **Chemistry** is the study of matter and its properties as well as how matter changes.
2. **Matter** is anything that has mass and takes up space.

Properties of Matter

- A **physical property** is a characteristic of a substance that can be observed without changing the substance's composition. Examples include:
 - _____ (e.g., color, shape, texture)
 - _____ (e.g., melting point, boiling point)
 - _____ (e.g., density, solubility)
- A **chemical property** is a characteristic of a substance that describes its ability to change into other substances. Examples include:
 - _____ (e.g., burning, rusting)

States of Matter

1. _____ - has a definite shape and volume.
Particles are tightly packed and vibrate in place.
2. _____ - has a definite volume but not a definite shape. Particles are close together but can move around.
3. _____ - has no definite shape or volume.
Particles are far apart and move around randomly.
4. _____ - exists at extremely high temperatures; particles are ionized.

Changes of Matter

- A **physical change** alters the form or appearance of a substance but does not change its composition. Examples include:
 - _____ (e.g., cutting, tearing)
 - _____ (e.g., melting, freezing, boiling)
 - _____ (e.g., dissolving)
- A **chemical change** results in the formation of new substances with different properties. Examples include:
 - _____ (e.g., burning, cooking)
 - _____ (e.g., rusting, tarnishing)

Elements and Compounds

Answers To The Chemistry Guided Work

Clifford Lane



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