

Lab 2 - Scientific Measurement

OBJECTIVES:

- Make quantitative and qualitative observations.
- Understand the importance of the Metric System (International System of Units) and how it is used in unit conversions.
- Identify and learn how to use standard laboratory tools used for measuring volume and mass.
- Utilize scientific method to investigate accuracy and precision of measuring tools.
- Calculate mean and standard deviation.
- Understand how polarity of various substances affect other substances in a mixture.
- Utilize the scientific method to investigate the different properties of water.

MATERIALS:

- Two decimal place laboratory balance
- 1 mL Transfer Pipet
- 10 mL Serological pipette and pump
- 10 mL Graduated cylinder
- 250 mL Beaker with deionized water
- 50 mL Beaker
- 2 Paperclips
- Petri dish
- 2 Glass slides
- Petroleum ether
- Vegetable oil
- Table Salt

SAFETY:

Understand and adhere to safe laboratory practices when performing any activity in the classroom or lab. Pay attention to any demonstrations of the protocol for the correct use of the instruments and materials necessary to complete the activities. Use personal protective equipment such as safety glasses or goggles, gloves, and aprons when appropriate. In this lab activity you will be using glassware, which may break if dropped, so please handle equipment with care.

BACKGROUND:

As a biology student, you will often be asked to make observations. These observations will be of either a qualitative or quantitative nature. A qualitative observation is a descriptive representation of a characteristic and is not numerical. Quantitative observations are numerical. Although qualitative observations can be valuable, quantitative observations are easier to evaluate and are preferred in laboratory science. For example, if you were studying rainfall in Florida and wanted to determine which city had the most rainfall last year, you could provide a quantitative description, e.g. 52 inches.

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