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LABORATORY MANUAL CHEMISTRY

(C.B.S.E. Syllabus)

Class XI



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Apc Chemistry Lab Manual Class 11

Brooks/Cole



Apc Chemistry Lab Manual Class 11:

Practical/Laboratory Manual Chemistry Class - XI Er. Meera Goyal, 2021-05-29

1 Basic Laboratory Techniques

1 To cut a glass tube or glass rod
2 To bend the glass rod at an angle
3 To draw a glass jet from a glass tube
4 To bore a cork and fit a glass tube into it
Viva Voce

2 Characterisation and Purification of Chemical Substances

1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique
Viva Voce
2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method
Viva Voce
3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample
4 To prepare the pure crystals of copper sulphate from the given crude sample
5 To prepare pure crystals of benzoic acid from a given impure sample
Viva Voce

3 Measurement of pH Values

1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper
2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration
3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper
4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH
5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+
Viva Voce

4 Chemical Equilibrium

1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions
2 To study the shift in equilibrium between $Co(H_2O)_6^{2+}$ and Cl^- ions by changing the concentrations of either of the ions
Viva Voce

5 Quantitative Analysis

1 To prepare M 10 oxalic acid solution by direct weighing method
2 To prepare M 10 solution of sodium carbonate by direct weighing method
3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid
4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution
Viva Voce

6 Qualitative Analysis

Analysis of Anions
Analysis of Cations
Viva Voce

7 Detection of Elements in Organic Compounds

1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test
2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test
Viva Voce

INVESTIGATORY PROJECTS

1 Checking of Bacterial Contamination in Water

1 To check the bacterial contamination in drinking water by testing sulphide ions
Viva Voce

2 Methods of Water Purification

1 To purify water from suspended impurities by using sedimentation
2 To purify water by boiling
3 To purify water by distillation method
4 To purify water by reverse osmosis technique
5 To purify water by GAC method
6 To purify water by bleach treatment
7 To purify water by oxidising agent
8 To purify water by ozone treatment method
Viva Voce

3 Water Analysis

1 To test the hardness of different water samples
Viva Voce

4 Foaming Capacity of Various Soaps

1 To compare the foaming capacity of different washing soaps
2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap
Viva Voce

5 Tea Analysis

1 To study the acidity of different samples of tea leaves tea by using pH paper
Viva Voce

6 Analysis of Fruits and Vegetable Juices

1 To analysis the fruit and vegetable juices for the constituent present in them

Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce Chemistry Lab Manual Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar, Lab Manual **Acp Chem 1172 General Chemistry Lab Manual** Brooks/Cole,2015-11-11 **Current Index to Journals in Education** ,1977 Acp Chem 1171 General Chemistry Lab Manual Brooks/Cole,2014-04-25 *The Software Encyclopedia* ,1986 Popular Photography ,1990-11 *Nutrition Abstracts and Reviews* ,1998 **General Chemistry I Laboratory Manual, 11E - CHM 111** Brightpoint Community College,2023-05-26 Literature on Information Retrieval and Machine Translation Service Bureau Corporation,1959

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In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Apc Chemistry Lab Manual Class 11**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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Table of Contents Apc Chemistry Lab Manual Class 11

1. Understanding the eBook Apc Chemistry Lab Manual Class 11
 - The Rise of Digital Reading Apc Chemistry Lab Manual Class 11
 - Advantages of eBooks Over Traditional Books
2. Identifying Apc Chemistry Lab Manual Class 11
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Apc Chemistry Lab Manual Class 11
 - User-Friendly Interface
4. Exploring eBook Recommendations from Apc Chemistry Lab Manual Class 11
 - Personalized Recommendations
 - Apc Chemistry Lab Manual Class 11 User Reviews and Ratings
 - Apc Chemistry Lab Manual Class 11 and Bestseller Lists

5. Accessing Apc Chemistry Lab Manual Class 11 Free and Paid eBooks
 - Apc Chemistry Lab Manual Class 11 Public Domain eBooks
 - Apc Chemistry Lab Manual Class 11 eBook Subscription Services
 - Apc Chemistry Lab Manual Class 11 Budget-Friendly Options
6. Navigating Apc Chemistry Lab Manual Class 11 eBook Formats
 - ePub, PDF, MOBI, and More
 - Apc Chemistry Lab Manual Class 11 Compatibility with Devices
 - Apc Chemistry Lab Manual Class 11 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Apc Chemistry Lab Manual Class 11
 - Highlighting and Note-Taking Apc Chemistry Lab Manual Class 11
 - Interactive Elements Apc Chemistry Lab Manual Class 11
8. Staying Engaged with Apc Chemistry Lab Manual Class 11
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Apc Chemistry Lab Manual Class 11
9. Balancing eBooks and Physical Books Apc Chemistry Lab Manual Class 11
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Apc Chemistry Lab Manual Class 11
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Apc Chemistry Lab Manual Class 11
 - Setting Reading Goals Apc Chemistry Lab Manual Class 11
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Apc Chemistry Lab Manual Class 11
 - Fact-Checking eBook Content of Apc Chemistry Lab Manual Class 11
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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