

Chapter 24

Budgetary Control and Responsibility Accounting

STUDY OBJECTIVES

After studying this chapter, you should be able to:

- 1 Describe the concept of budgetary control.
- 2 Evaluate the usefulness of static budget reports.
- 3 Explain the development of flexible budgets and the usefulness of flexible budget reports.
- 4 Describe the concept of responsibility accounting.
- 5 Indicate the features of responsibility reports for cost centers.
- 6 Identify the content of responsibility reports for profit centers.
- 7 Explain the basis and formula used in evaluating performance in investment centers.



The Navigator

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Answer Self-Study Questions	
Complete Assignments	

Feature Story

TRYING TO AVOID AN ELECTRIC SHOCK

Budgets are critical to evaluating an organization's success. They are based on management's expectations of what is most likely to happen in the future. In order to be useful, they must be accurate. But what if management's expectations are wrong? Estimates are never exactly correct, and

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COLIN M. DRURY



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$$\begin{aligned} & \text{N}_2 + 3 \text{H}_2 \rightarrow 2 \text{NH}_3 \\ & 2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2 \\ & 2 \text{NaCl} + \text{F}_2 \rightarrow 2 \text{NaF} + \text{Cl}_2 \\ & 2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O} \\ & \text{Pb} + \text{OH}^- + 2 \text{HCl} \rightarrow 2 \text{H}_2\text{O} + 1 \text{PbCl}_2 \\ & 2 \text{Al} + 3 \text{Br}_2 \rightarrow 2 \text{AlBr}_3 \\ & 3 \text{K}_2\text{SO}_4 \rightarrow 6 \text{KBr} + 1 \text{Al}_2\text{SO}_4 + 3 \text{CH}_4 + 2 \text{O}_2 \\ & \text{CO}_2 + 2 \text{H}_2\text{O} \rightarrow \text{C}_3\text{H}_8 + 5 \text{O}_2 \\ & 3 \text{CO}_2 + 4 \text{H}_2\text{O} \rightarrow \text{C}_8\text{H}_{18} + 25 \text{O}_2 \\ & 16 \text{CO}_2 + 18 \text{H}_2\text{O} \end{aligned}$$

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gaseous butane C_4H_{10} reacts with diatomic oxygen gas to yield gaseous carbon dioxide and water vapor

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$$\begin{aligned} & \text{Na} + 3 \text{PO}_4 + 3 \text{KOH} + 3 \text{NaOH} + 1 \text{K}_3\text{PO}_4 + 1 \text{MgF}_2 + 1 \text{Li}_2\text{CO}_3 + 1 \text{MgCO}_3 + 2 \text{LiF} + 1 \text{P}_4\text{O}_{10} + 2 \text{P}_2\text{O}_5 + 3 \text{RbNO}_3 + 1 \text{BeF}_2 + 1 \text{BeNO}_3 + 2 \text{RbF} \end{aligned}$$

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$$\begin{aligned} & \text{N}_2 + 3 \text{H}_2 \rightarrow 2 \text{NH}_3 \\ & \text{HNO}_3 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{O} + \text{O}_2 + 3 \text{H}_2\text{O} \\ & 2 \text{H}_2\text{O} + 2 \text{O}_2 + 4 \text{RbH} + 2 \text{O} + \text{RbOH} + 2 \text{H}_2\text{O} + 5 \text{Cl}_2 + 2 \text{KBr} + \text{KCl} + \text{Br}_2 + 6 \text{N}_2 + 2 \text{H}_2 + 2 \text{NH}_3 + 7 \text{N}_2\text{F}_6 + 3 \text{Ca} + \text{N}_2\text{O} + \text{C}_2\text{H}_5\text{OH} \end{aligned}$$

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