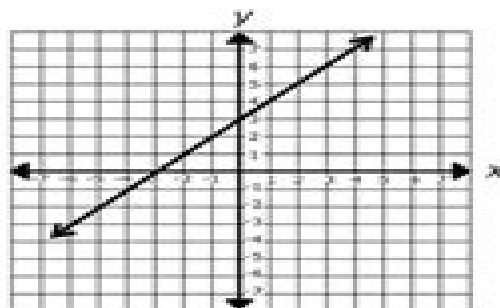


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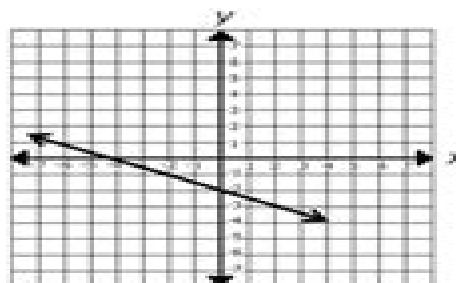
1



Which is most likely the equation of the line shown on the graph above?

- A $y = 3x$
- B $y = x + 3$
- C $y = x - 3$
- D $y = 3x + 1$

2



Which is most likely the equation of the line shown on the graph above?

- A $y = 2x - 2$
- B $y = -2x - \frac{1}{2}$
- C $y = -\frac{1}{2}x - 2$
- D $y = -2x - 1$

Apex Algebra 1 Unit 5 Answers

Paul Procter



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web this data shows that there was a gain the mass of the lighter which shows that there was a certain amount of butane 0.38 grams 0.02 grams gas that was successfully collected during the gas collection there is a mild influence of barometric pressure during both trials as shown above in the raw data

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web 11 use avogadro's law to determine the number of moles of butane gas assume that butane is an ideal gas and that one mole has a volume of 22.4 l at stp $v_1/n_1 = v_2/n_2$ $0.0896 \text{ l} / n_2 = 22.4 \text{ l} / 1 \text{ mole}$ $n_2 = 0.0040 \text{ moles}$ 12 experimental molar mass g mole of butane 0.23 g / 0.0040 moles 58 g mole 13 the molecular formula of butane is C_4H_{10}

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