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web there s 2 on the left and 1 on the right so we need to change the coefficient of NH_3 to 2 now we have $\text{N}_2 + \text{H}_2 \rightarrow 2 \text{ NH}_3$ total

the atoms up again on the left there is 2 n and 2 h still on the right there is 2 n and 6 h now so now all we need to do is make the left side have 6 h in total

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web in order to balance a chemical equation the quantities of each type of element and polyatomic ion that are present in the reactants and the products of the reaction must be determined the first reactant in this equation contains magnesium mg and a hydroxide ion oh⁻ which is a polyatomic anion

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