

Assigning Oxidation Numbers

Rule	Example
1. The oxidation number of any uncombined element is 0.	The oxidation number of $\text{Na}(s)$ is 0.
2. The oxidation number of a monatomic ion equals the charge on the ion.	The oxidation number of Cl^- is -1.
3. The more-electronegative element in a binary compound is assigned the number equal to the charge it would have if it were an ion.	The oxidation number of O in NO is -2.
4. The oxidation number of fluorine in a compound is always -1.	The oxidation number of F in LiF is -1.
5. Oxygen has an oxidation number of -2 unless it is combined with F, in which it is +1 or +2, or it is in a peroxide, in which it is -1.	The oxidation number of O in NO_2 is -2.
6. Hydrogen's oxidation state in most of its compounds is +1 unless it is combined with a metal, in which case it is -1.	The oxidation number of H in LiH is -1.
7. In compounds, the elements of Groups 1 and 2 as well as aluminum have oxidation numbers of +1, +2, and +3, respectively.	The oxidation number of Ca in CaCO_3 is +2.
8. The sum of the oxidation numbers of all atoms in a neutral compound is 0.	The oxidation number of C in CaCO_3 is +4.
9. The sum of the oxidation numbers of all atoms in a polyatomic ion equals the charge of the ion.	The oxidation number of P in H_2PO_4^- is +5.

Assigning Oxidation Numbers Answers

SJ Ball



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The book delves into Assigning Oxidation Numbers Answers. Assigning Oxidation Numbers Answers is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Assigning Oxidation Numbers Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Assigning Oxidation Numbers Answers
 - Chapter 2: Essential Elements of Assigning Oxidation Numbers Answers
 - Chapter 3: Assigning Oxidation Numbers Answers in Everyday Life
 - Chapter 4: Assigning Oxidation Numbers Answers in Specific Contexts
 - Chapter 5: Conclusion
2. In chapter 1, the author will provide an overview of Assigning Oxidation Numbers Answers. The first chapter will explore what Assigning Oxidation Numbers Answers is, why Assigning Oxidation Numbers Answers is vital, and how to effectively learn about Assigning Oxidation Numbers Answers.
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4. In chapter 3, this book will examine the practical applications of Assigning Oxidation Numbers Answers in daily life. The third chapter will showcase real-world examples of how Assigning Oxidation Numbers Answers can be effectively utilized in everyday scenarios.
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6. In chapter 5, the author will draw a conclusion about Assigning Oxidation Numbers Answers. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Assigning Oxidation Numbers Answers.

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