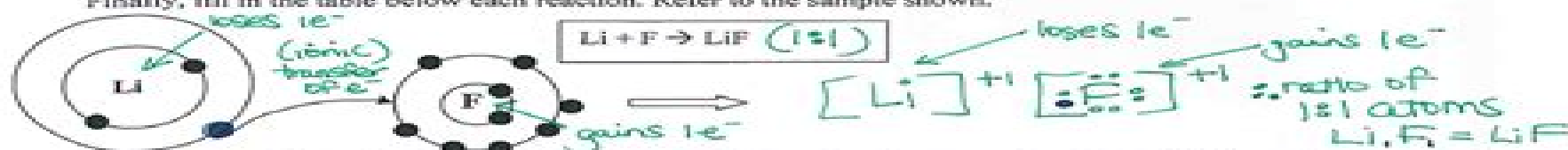


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Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1 ve	transfers $1e^-$ to F	Li^{+}
F	7 ve	accepts $1e^-$ from Li	F^{-}

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $\text{Li} + \text{Cl} \Rightarrow \text{LiCl}$	Li	1 ve	donates $1e^-$	Li^{+1}
$\text{Li} + \text{Cl} \rightarrow [\text{Li}]^{+1} [\text{Cl}]^{-1}$	Cl	1 ve	accepts $1e^-$	Cl^{-1}
2) $\text{Ca} + \text{O} \Rightarrow \text{CaO}$	Ca	2 ve	donates $2e^-$	Ca^{+2}
$\text{Ca} + \text{O} \rightarrow [\text{Ca}]^{+2} [\text{O}]^{-2}$	O	6 ve	accepts $2e^-$	O^{-2}
3) $\text{Be} + \text{F} \Rightarrow \text{BeF}_2$	Be	2 ve	donates $2e^-$ to each F	Be^{+2}
$\text{Be} + \text{F} \rightarrow [\text{Be}]^{+2} [\text{F}]^{-1}$	F	7 ve	each F accepts $1e^-$ each	F^{-1}
4) $\text{Mg} + \text{S} \Rightarrow \text{MgS}$	Mg	2 ve	donates $2e^-$ to S	Mg^{+2}
$\text{Mg} + \text{S} \rightarrow [\text{Mg}]^{+2} [\text{S}]^{-2}$	S	6 ve	accepts $2e^-$	S^{-2}
5) $\text{K} + \text{F} \Rightarrow \text{KF}$	K	1 ve	donates $1e^-$	K^{+1}
$\text{K} + \text{F} \rightarrow [\text{K}]^{+1} [\text{F}]^{-1}$	F	7 ve	accepts $1e^-$	F^{-1}

Bond Formation Study Guide Answers

David R. Klein



Bond Formation Study Guide Answers:

Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems

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Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

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scientists can benefit from improved critical thinking and problem solving skills that are developed from the study of organic chemistry Organic chemistry like any skill is best learned by doing It is difficult to learn by rote memorization and true understanding comes only from concentrated reading and working as many problems as possible In fact problem sets are the best way to ensure that concepts are not only well understood but can also be applied to real world problems in the work place Helps readers learn to categorize analyze and solve organic chemistry problems at all levels of difficulty Hundreds of fully worked practice problems all with solutions Key concept summaries for every chapter reinforces core content from the companion book Study Guide and Solutions Manual Neil E. Schore,2002-08-02 **Organic Chemistry, 5e Student**

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Study Guide & Student Solutions Manual T. W. Graham Solomons,Craig B. Fryhle,Scott A. Snyder,2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder s tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional

groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

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