Acid-Base Bafance

O-6.9

Acidic

Alkaline

pH of 7 = Neutral (Water)

pH of Blood	Kanges 7,35-7.45
ETCO2	35-45mmHg
HCO3 (Bicarbonate)	22-26 meq/L

HCO3: Normally, O2 combines with glucose to create energy and CO2 is an acid waste by product of this. Bicarbonate is a buffer produced by the body that is alkaline, and binds with CO2 to neutralize it to allow the body to get rid of it.

Acid Base Study Guide Answer Key

Karen Timberlake

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