

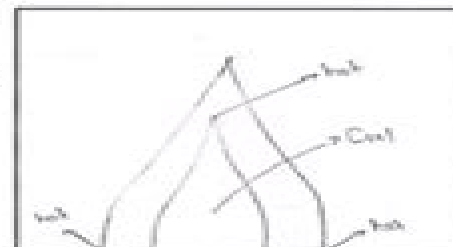
Experiment 1 Report Sheet

Basic Laboratory Operations

A. Bunsen Burner

1. Instructor's Approval of a well-adjusted Bunsen flame: _____
2. a. At right, draw a sketch of the heat zones for a nonluminous flame as directed with the wire gauze positioned parallel to the burner barrel. Label the "cool" and "hot" zones.
 - b. What happens to the "cool" and "hot" zones of a nonluminous flame, when the air-control valve is closed (nearly closed)? Explain.

As the air-control valve approaches the nearly closed position, the "cool" and "hot" zones merge into one flame.



B. Laboratory Balances

1. Determine the mass of the following objects on your assigned balance. Express your results with the correct sensitivity.

Balance no. _____ Sensitivity 0.001 g

Object	Description of Object (size or volume)	Mass (g)
Test tube		<u>5.240 g</u>
Beaker	<u>50 mL</u>	<u>29.052 g</u>
Spatula		<u>7.401 g</u>
Graduated cylinder	<u>10 mL</u>	<u>15.711 g</u>
Other (see instructor)	<u>Unknown #2</u>	<u>2.135 g</u>

2. Precision of Measurement. Balance no. _____

Mass of graduated cylinder (if not tared) 15.711 g

	Trial 1	Trial 2	Trial 3	Trial 4	Trial 5	Trial 6
a. Mass of cylinder + H ₂ O (l) (g)	<u>20.261 g</u>	<u>25.423 g</u>	<u>31.002 g</u>			
b. Mass of 7 mL H ₂ O (l) (g)	<u>4.55 g</u>	<u>4.71 g</u>	<u>15.511 g</u>			
c. Average mass of 7 mL H ₂ O (l) (g)			<u>4.75 g</u>			

Comment on the precision of the mass measurements for the water samples.

Trial #3 does not seem to be precise.

Date Analysis, B

Beran Laboratory Manual Experiment 2 Report Sheet

Arthur D. Baker



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1. Understanding the eBook Beran Laboratory Manual Experiment 2 Report Sheet
 - The Rise of Digital Reading Beran Laboratory Manual Experiment 2 Report Sheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Beran Laboratory Manual Experiment 2 Report Sheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Beran Laboratory Manual Experiment 2 Report Sheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Beran Laboratory Manual Experiment 2 Report Sheet
 - Personalized Recommendations
 - Beran Laboratory Manual Experiment 2 Report Sheet User Reviews and Ratings
 - Beran Laboratory Manual Experiment 2 Report Sheet and Bestseller Lists
5. Accessing Beran Laboratory Manual Experiment 2 Report Sheet Free and Paid eBooks
 - Beran Laboratory Manual Experiment 2 Report Sheet Public Domain eBooks
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 - Beran Laboratory Manual Experiment 2 Report Sheet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Beran Laboratory Manual Experiment 2 Report Sheet
 - Highlighting and Note-Taking Beran Laboratory Manual Experiment 2 Report Sheet
 - Interactive Elements Beran Laboratory Manual Experiment 2 Report Sheet
8. Staying Engaged with Beran Laboratory Manual Experiment 2 Report Sheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Beran Laboratory Manual Experiment 2 Report Sheet
9. Balancing eBooks and Physical Books Beran Laboratory Manual Experiment 2 Report Sheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Beran Laboratory Manual Experiment 2 Report Sheet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Beran Laboratory Manual Experiment 2 Report Sheet
 - Setting Reading Goals Beran Laboratory Manual Experiment 2 Report Sheet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Beran Laboratory Manual Experiment 2 Report Sheet
 - Fact-Checking eBook Content of Beran Laboratory Manual Experiment 2 Report Sheet
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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