

### **Pre-lab 3 | Analysis of Antacids by Acid-Base Titration**

#### **Materials:**

1. Commercial Antacid Tablets (Two different brands)
2. Mortar and Pestle
3. 125 mL Erlenmeyer Flasks (x2)
4. 25 mL Pipette
5. Hot Plate
6. Bromophenol Blue Indicator
7. ~0.1 M Hydrochloric Acid (HCl) Solution
8. Sodium Hydroxide (NaOH) Solution
9. Burette
10. Funnel
11. Distilled Water

#### **Procedures:**

1. Prepare the Sample:
  - Obtain two different kinds of commercial antacid tablets.
  - Keep track of the brand name, tablet count, bottle cost, and active ingredient mass per tablet.
  - One pill should be ground into a fine powder and weighed at about 0.2g.
2. HCl Solution Preparation
  - Pipette 25.0 mL of standardized 0.1 M HCl into each flask containing powdered antacid.
  - Take careful note of the HCl solution's exact concentration.
3. Removal of CO<sub>2</sub>:
  - Gently heat the flasks for about 1 minute with intermittent swirling to remove any dissolved CO<sub>2</sub>.
  - Add bromophenol blue indicator and check if the solution is yellow (indicating adequate HCl).
4. Standardized NaOH Solution:
  - Collect approximately 50 mL of standardized NaOH solution and record its concentration.
5. Titration:
  - Allow the antacid solution to cool to room temperature before titrating.
  - Titrate the antacid sample with NaOH solution until it reaches a pale blue terminus.
  - Keep track of the beginning and final quantities of NaOH solution.
6. Repeat for Second Antacid:
  - For the second brand of antacid tablet, repeat steps 1-5.
7. Cleaning Up:
  - Unplug and turn off the hot plate.

# Analysis Of Antacids Manual

**Beran**



## Analysis Of Antacids Manual:

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