

10.1.1. The student will be able to identify the reagents and conditions required for the substitution of a halogenoalkane with a nucleophile. The student will be able to identify the reagents and conditions required for the substitution of a halogenoalkane with a nucleophile.

$$\text{R-X} + \text{Nu}^- \rightarrow \text{R-Nu} + \text{X}^- \quad (10.1)$$

where R is the halogenoalkane, Nu⁻ is the nucleophile, R-Nu is the substitution product, X⁻ is the halide ion. The nucleophile is a species with a lone pair of electrons and a negative charge. The nucleophile attacks the carbon atom bonded to the halogen atom, forming a new bond. The halide ion is a species with a lone pair of electrons and a negative charge.

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Summary of Learning Objectives

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Padhraic Smyth



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