Ponic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms Valence electrons Electron transfer from/to each atom
Li Ive Housters Ive to F.
F Tye accepts Ive from Li F

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	loss formed in the product
Li +ci→ Lici Li +ci→ Lici Li +ci→ Lici	Li	Ive	donate le-	Lit
	CI	Ive	accepts 1e	C1-1
2) Ca+0=> Ca0 (a) + 1 - 0: -> [a] +2 [0:] 2	2 Ca	Dre	Starch 205	Cata
	0	6ve	accepts	0-
Be F: BeF:	-1 Re	Que	donates de to	Beto
	F	712	each F accepts Ive each	
mg + 5 → [mg]+2[:S:]2	2 mg	Sve	denotes ae to	wet2
	S	Gve	assepts a e	5->
S)K+F⇒KF KU+FE: → [K]+[E]	K	Ivē	donates	K+I
	=	7ve	accepts	F-/

Bond Formation Study Guide Answers

T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder

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groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

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