4

Atomic Structure

Reviewsheet

A. Completion

Use this completion exercise to check your understooding of the concepts and terms that are introduced in this chapter. Each black can be completed with a term, shore phoses, or number.

It was ___7__ who discovered the nucleus of the atom. The nucleus has a ___8__ charge and it occupies a very small volume of the atom. In contrast, the negatively charged ___9__ occupy most of the volume of the atom.

The number of __19_ in the nucleus of the atom is the atomic __11_ of that element: Because atoms are electrically neutral, the number of protons and __12_ in an atom are equal. The sum of the __13_ and neutrons is the mass number. Atoms of the same element are identical in most respects, but they can differ in the number of __14_ in the nucleus. Atoms that have the same number of protons but different mass numbers are called __15__.

The ___16__ of an element is the weighted average of the __2 masses of the isotopes of that element. Two isotopes of sulfur are __12 S and __14 S. An atom of the sulfur—32 isotope contains __17__ protons and __18__ neutrons. The sulfur—34 isotope has __19__ protons and __20__ neutrons.

different-	4.2
sob aboute	4.2
repative	4.2
6x01003	4.2
1360 HOWE	4.2
nucleus .	4.9
Rutnertore	6.5
positive	
electrans	4.8
prodons (elec)	14
ramber	4.4
evectore	4.4
newtrons	
isotopes	4.6
16	4.5
	4.6
	4.6
18	4.5
	sub etamic regative protons regative protons etectrons protons (elec)

Atomic Structure Answer Key Study Guide

Marcia L. Gillette,Ira Blei,Wendy Gloffke,George Odian

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