1. Define the terms acid and base using the Arrhenius and Bronsted/Lowry definitions.

	Arrhenius	Bronsted/Lowry	
Acid			
Base			

Acids and bases share some properties but are different in others. Using your knowledge of acids and bases, fill in the column in the table below.

Property	Acids	Bases
Conductivity in Aqueous Solution		
Taste		
Feel		
Reaction with Metal		
Reaction with Carbonate	ĺ	
Color with Litmus Paper		
Color with Universal Indicator		
Color with Phenolphthalein		

- Using your knowledge of the Brønsted-Lowry theory of acids and bases, complete
 the equations for the following acid-base reactions and indicate each conjugate
 acid-base pair.
 - a) HNO, + OH: →
 - b) CH₃NH₂ + H₂O →

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