Acid-Base Bafance

O-6.9

Acidic

Alkaline

pH of 7 = Neutral (Water)

| pH of Blood | Kanges 7.35-7.45 |
|-----------------------|---------------------|
| ETCO2 | 35-45mmHg |
| HCO3 (Bicarbonate) | 22-26 meq/L |

HCO3: Normally, O2 combines with glucose to create energy and CO2 is an acid waste by product of this. Bicarbonate is a buffer produced by the body that is alkaline, and binds with CO2 to neutralize it to allow the body to get rid of it.

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