# Table 18.2

Salt	K <sub>sp</sub>	Salt	K <sub>sp</sub>	Salt	K <sub>sp</sub>
Halides	0,50	Sulfates	250	Hydroxides	7780
AgCI	$1.8 \times 10^{-10}$	PbSO <sub>4</sub>	6.3 × 10 <sup>-7</sup>	Al(OH) <sub>3</sub>	$3.0 \times 10^{-34}$
AgBr	$5.0 \times 10^{-13}$	BaSO,	$1.1 \times 10^{-19}$	Zn(OH) <sub>2</sub>	$3.0 \times 10^{-16}$
Agl	$8.3 \times 10^{-17}$	CaSO <sub>4</sub>	$2.4 \times 10^{-6}$	Ca(OH) <sub>2</sub>	$6.5 \times 10^{-4}$
PbCI <sub>2</sub>	1.7 × 10 <sup>-5</sup>	Sulfides	1 Description	Mg(OH) <sub>2</sub>	$7.1 \times 10^{-12}$
PbBr <sub>2</sub>	$2.1 \times 10^{-6}$	NiS	$4.0 \times 10^{-29}$	Fe(OH) <sub>2</sub>	$7.9 \times 10^{-16}$
Pbl <sub>2</sub>	$7.9 \times 10^{-9}$	CuS	$8.0 \times 10^{-37}$	Carbonates	
PbF <sub>3</sub>	$3.6 \times 10^{-6}$	Ag <sub>7</sub> S	$8.0 \times 10^{-61}$	CaCO <sub>3</sub>	$4.5 \times 10^{-9}$
CaF <sub>2</sub>	$3.9 \times 10^{-11}$	ZnS	$3.0 \times 10^{-23}$	SrCO <sub>3</sub>	$9.3 \times 10^{-10}$
Chromates	novel description	FeS	$8.0 \times 10^{-19}$	ZnCO <sub>3</sub>	$1.0 \times 10^{-10}$
PbCrO <sub>x</sub>	$1.8 \times 10^{-14}$	CdS	$1.0 \times 10^{-37}$	Ag <sub>2</sub> CO <sub>3</sub>	$8.1 \times 10^{-12}$
Ag <sub>2</sub> CrO <sub>4</sub>	$1.2 \times 10^{-12}$	PbS	$3.0 \times 10^{-38}$	BaCO <sub>a</sub>	$5.0 \times 10^{-9}$



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Michael Dzwiniel,Frank Jenkins,Dick Tompkins,Hans Van Kessel

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