

Additivity of Heats of Reaction: Hess's Law

In this experiment you will use a Styrofoam cup calorimeter to measure the heat released by three reactions/physical changes. One of them is the same as the sum of the other two. Therefore, according to Hess's Law of Constant Heat Summation (or just Hess's Law), the heat of reaction of the one reaction should be equal to the sum of the heats of reaction for the other two. The primary objective of this experiment is to confirm this law.

The reactions we will use in this experiment are:

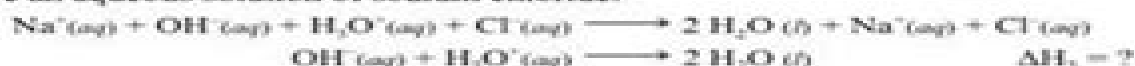
- (1) Solid sodium hydroxide dissolves in water to form an aqueous solution of ions.



- (2) Solid sodium hydroxide reacts with an aqueous solution of hydrochloric acid to form water and leave an aqueous solution of sodium chloride.



- (3) Aqueous solutions of sodium hydroxide and hydrochloric acid react to form water and leave an aqueous solution of sodium chloride.



You will use a Styrofoam cup (in a beaker) as a calorimeter, as shown in Figure 1. For purposes of this experiment, you may assume that the heat loss to the calorimeter and the surrounding air is negligible. Even if heat is lost to either of these it is a fairly constant factor in each part of the experiment and has little effect on the final results.



Figure 1

Additivity Of Heats Of Reaction Lab Answers

R Sanford



Additivity Of Heats Of Reaction Lab Answers:

The Enigmatic Realm of **Additivity Of Heats Of Reaction Lab Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Additivity Of Heats Of Reaction Lab Answers** a literary masterpiece penned by a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those that partake in its reading experience.

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