



Buffers in Household Products

Concepts:

Buffer, weak acids and weak bases, conjugate acid–base pair, dissociation constant, neutralization, titration

Use the following recommendations to increase and/or decrease the challenge difficulty for your students.

Short-on-time Inquiry Lab:

First, students are assigned a household product for testing. Then students follow a detailed procedure to titrate the product. Lastly, the data are graphed and analyzed. Students learn about titration curves and buffering capability.

Guided Inquiry Lab:

First, students titrate an acid in the introductory activity. Then students follow a detailed procedure to titrate an assigned household product. Lastly, the data are graphed and analyzed. Students learn about titration curves and buffering capability.

Open Inquiry Lab:

First, students titrate an acid in the introductory activity. Then after being assigned a household product to test, students design a procedure to analyze the product's buffering ability. Students graph multiple titration curves and analyze their results.

Advanced Inquiry Lab:

This lab builds on the open version. In addition to activities in the open procedure, students analyze a second household product and research biological applications of buffering systems.

Outcomes:

Students will become familiar with buffers and titration curves. They will better understand what components are needed in a buffer system.

Associated Phenomena:

Buffers, Acids and Bases

Standards

Science & Engineering Practices	Disciplinary Core Ideas	Crosscutting Concepts
Constructing explanations and designing solutions	HS-PS1.B: Chemical Reactions	Patterns

Performance Expectations

HS-PS1-2: Construct and revise an explanation for the outcome of a simple chemical reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns of chemical properties.

Buffers In Household Products Flinn Answers

ML Morrison



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