

Chemistry of the Elements

Lecture 3B – s block Group 2

1st Year UG course – Term 2

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Group 2 – Alkaline Earth Metals

Beryllium

[He] $2s^2$

Magnesium

[Ne] $3s^2$

Calcium

[Ar] $4s^2$

Strontium

[Kr] $5s^2$

Barium

[Xe] $6s^2$

Radium

[Xe] $7s^2$

- The elements in group 2 are metallic solids at ambient temperature and pressure
- All the isotopes of radium (Ra) are radioactive; ^{226}Ra is the longest lived with $t_{1/2} = 1599$ years
- Chemistry is dominated by the M^{2+} cations

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