

General Biology Final Exam Study Guide

The formation of a peptide bond is the result of which type of reaction?

- A. condensation reaction
- B. nucleotide reaction
- C. hydrolysis reaction
- D. nuclear reaction -   -A

Which of the levels of protein organization is correctly paired with the interaction that might be responsible for producing the structure?

- A. primary structure; ionic bonding between different amino acids
- B. secondary structure; hydrogen bonding between the R-groups (side chains) of different amino acids
- C. tertiary structure; van der Waals interactions between hydrophobic amino acids
- D. quaternary structure; disulfide bonds between different cysteine residues in the same polypeptide -   -C

A condensation reaction produces a phosphodiester bond between the 3' carbon of the sugar of one nucleotide and the phosphate of the next nucleotide.

- A. True
- B. False -   -A

Which of the following would be found in both a ribonucleotide and a deoxyribonucleotide?

- A. uracil
- B. thymine
- C. OH on the 2' carbon on the sugar
- D. OH on the 3' carbon on the sugar -   -D (sugar is ribose or deoxyribose on the 2' OH)

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web 33mol o₂ 6mol br₂ 11mol o₂ 18mol br₂ identify the limiting reactant s and excess reactant s the limiting reactant is o₂ since it would yield the least amount of product 18 mol br₂ the excess reactant is c₂h₃br₃ since its complete reaction would have yielded up to 54 mol br₂ think about your result

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web step 4 the reactant that produces a smaller amount of product is the limiting reagent mg mg produces less mgo mgo than does o₂ o₂ 3 98 g mgo vs 25 2 g mgo therefore mg is the limiting reagent in this reaction step 5 the reactant that produces a larger amount of product is the excess reagent

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web sep 15 2023 introduction understanding limiting and excess reactants is key in chemistry pogil is an interactive approach that helps students grasp these concepts it provides guided questions and activities here we ll explore pogil answers for limiting and excess reactants and their role in predicting reactions

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web please save your changes before editing any questions 5 minutes 1 pt viewing model 3 if 4 moles of hydrogen react with 2 moles of oxygen which of these is the excess reactant is leftover at the end and how many moles of it are leftover hydrogen

2 moles leftover hydrogen 1 mole leftover

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web if you start with 14 82 g of Ca(OH)_2 and 16 35 g of H_2SO_4 a determine the limiting reagent b determine the number of moles of H_2O produced c determine the number of grams of CaSO_4 produced d determine the number of grams of excess reagent left 1 make sure the equation is balanced

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