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 $3\text{Ca(OH)}_2 + 1\text{Al}_2\text{SO}_4 \rightarrow 3\text{CaSO}_4 + 2\text{Al(OH)}_3$ type of reaction double displacement
 $3\text{Mg} + 1\text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$ type of reaction single displacement
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 $2\text{Fe} + 3\text{MgO} \rightarrow 3\text{Mg} + \text{Fe}_2\text{O}_3$ type of reaction double displacement
 $4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$ type of reaction single displacement
 $1\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2$

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 $\text{NaOH} + \text{KNO}_3 \rightarrow \text{NaNO}_3 + \text{KOH}$ type of reaction double displacement
 $\text{CH}_2\text{O} + \text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_2\text{H}_4 + 2\text{O}_2$ type of reaction double displacement
 $2\text{Fe} + 6\text{NaBr} \rightarrow 2\text{FeBr}_3 + 3\text{Na}_2\text{O}$ type of reaction double displacement
 $\text{MgSO}_4 + 2\text{NH}_4\text{OH} \rightarrow \text{Mg(OH)}_2 + (\text{NH}_4)_2\text{SO}_4$ type of reaction double displacement
 $\text{PbO}_2 + 2\text{Na}_2\text{CO}_3 \rightarrow \text{PbO} + 2\text{Na}_2\text{O} + \text{CO}_2$ type of reaction double displacement
 $4\text{H}_2\text{O} + 8\text{FeO} + 4\text{CO}_2 \rightarrow 2\text{H}_2\text{O} + 4\text{HCl} + \text{NaOH} + \text{H}_2\text{O} + \text{NaCl}$

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 $4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$ type of reaction single displacement
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